

A triclinic polymorph of 1,4-dibenzoylbutane

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Key indicators

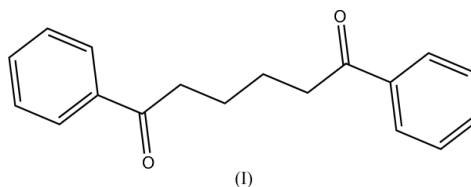
Single-crystal X-ray study
 $T = 173\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.036
 wR factor = 0.096
Data-to-parameter ratio = 17.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

A new polymorph of 1,4-dibenzoylbutane, $\text{C}_{18}\text{H}_{18}\text{O}_2$, has been discovered. Until now only a monoclinic polymorph of the title compound was known. The newly found triclinic polymorph crystallizes in $P\bar{1}$ with half a molecule in the asymmetric unit. The two polymorphs differ slightly in the torsion angles of the central chain.

Received 4 May 2004
Accepted 11 May 2004
Online 15 May 2004

Comment

The aim of our investigations (Ton, 2004) was to synthesize a compound with two hydrazone fragments in order to examine the intramolecular hydrogen-bond interactions in the solid state. Unfortunately, we obtained crystals of 1,4-dibenzoylbutane, (I), which was one of the starting materials. The structure of (I) has already been determined at room temperature in the monoclinic space group $P2_1/n$ with $Z' = 1$ (Deguire & Brisse, 1988). We collected the data set at low temperature and obtained, surprisingly, the space group $P\bar{1}$ with $Z' = \frac{1}{2}$. To verify that the crystal structure had not undergone a phase transition upon cooling, the same crystal was investigated again, at room temperature. We still obtained the space group $P\bar{1}$. Furthermore, we determined the structure at room temperature using a crystal which had not been cooled already, and again obtained the triclinic structure. In this way, a new polymorph of (I) has been discovered.



The molecule lies on a centre of symmetry, with one-half molecule in the asymmetric unit. Whereas (I) is essentially planar (the r.m.s. deviation of all the non-H atoms is 0.019 \AA), the monoclinic polymorph has an r.m.s. deviation of 0.255 \AA . The two polymorphs differ mainly in the torsion angles of the central chain. A least-squares fit of all non-H atoms of the two polymorphs yields an r.m.s. deviation of 0.114 \AA .

Experimental

1,4-Dibenzoylbutane was synthesized according to the method of Hünig *et al.* (1979). The intention was to use it in a further acid-catalysed reaction with benzhydrazide. After work-up, and following crystallization attempts in diethyl ether, block-shaped crystals were obtained.

Crystal data

$C_{18}H_{18}O_2$
 $M_r = 266.32$
 Triclinic, $P\bar{1}$
 $a = 5.7482$ (12) Å
 $b = 7.6480$ (16) Å
 $c = 8.2817$ (15) Å
 $\alpha = 95.413$ (16)°
 $\beta = 105.265$ (15)°
 $\gamma = 95.206$ (17)°
 $V = 347.13$ (12) Å³

$Z = 1$
 $D_x = 1.274$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 4194 reflections
 $\theta = 3.8$ – 27.7 °
 $\mu = 0.08$ mm⁻¹
 $T = 173$ (2) K
 Block, colourless
 0.52 × 0.29 × 0.27 mm

Data collection

Stoe IPDS-II two-circle diffractometer
 ω scans
 Absorption correction: none
 5151 measured reflections
 1610 independent reflections

1203 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.040$
 $\theta_{max} = 27.8$ °
 $h = -7 \rightarrow 7$
 $k = -9 \rightarrow 9$
 $l = -10 \rightarrow 10$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.036$
 $wR(F^2) = 0.096$
 $S = 1.08$
 1610 reflections
 91 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0517P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.29$ e Å⁻³
 $\Delta\rho_{min} = -0.17$ e Å⁻³

H atoms were included with fixed individual displacement parameters [$U_{iso}(H) = 1.2U_{eq}(C)$] using a riding model, with aromatic C–H = 0.95 Å and methylene C–H = 0.99 Å.

Data collection: *X-AREA* (Stoe & Cie, 2001); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine

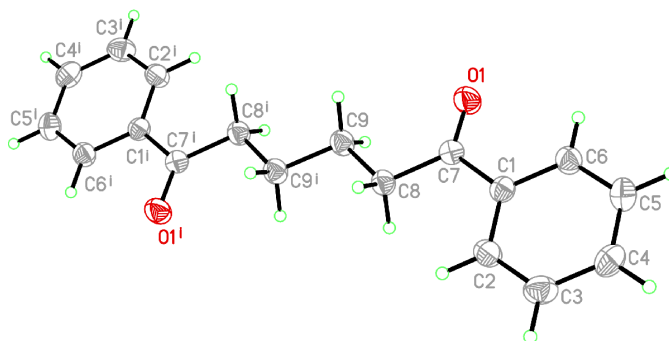


Figure 1
 Perspective view of the title compound, with the atom-numbering scheme; displacement ellipsoids are drawn at the 50% probability level. [Symmetry code: (i) $2 - x, 1 - y, 2 - z$.]

structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

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